Non Sibi High School

Andover's Chem 300: Accelerated/Honors Chemistry

Chapter 2, Review Quiz 1

1

Convert the following:

- a. 0.040 milligrams neon to number of neon atoms
- b. 7.43×10^{25} magnesium atoms to kilograms magnesium

$\mathbf{2}$

The density of liquid $C_2H_4(NH_2)_2$ is 0.90 g/mL. What is the volume in liters of 1.08×10^{26} molecules of $C_2H_4(NH_2)_2$?

3

The density of solid $C_{12}H_{22}O_{11}$ is 1.59 g/cm³. How many molecules are in 75 cm³ of $C_{12}H_{22}O_{11}$? How many carbon atoms are in this sample?



This work is licensed under a <u>Creative Commons Attribution-NonCommercial-NoDerivs 3.0 Unported License</u> <u>Contact: kcardozo@andover.edu</u>