Non Sibi High School

Andover's Chem 550/580: Advanced Chemistry

Chapter 9, Review Quiz 1

1

Write the chemical formula for each compound:

- a. ammonium phosphate
- b. butanol
- c. aluminum sulfate
- d. potassium nitride

$\mathbf{2}$

Write the name of each compound:

a. Fe(OH)₃ b. P₂O₅ c. Mg(HCO₃)₂ d. C₆H₁₄

3

Draw Lewis structures for:

a. boron trichloride b. SF_5 ⁻

$\mathbf{4}$

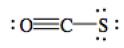
Draw Lewis structures for H_2CNH and CH_3NH_2 . Which molecule will have the longer carbon-to-nitrogen bond length?

$\mathbf{5}$

Draw all the resonance structures for NO $_2$ $^-$

6

Show any non-zero formal charges on the following Lewis structure:





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