Non Sibi High School

Andover's Chem 250: Introductory/Basic Chemistry

Chapter 12, Review Quiz 1

1

Rank the compounds $CaBr_2$, KI, MgS, and $SiCl_4$ from lowest to highest melting point.

$\mathbf{2}$

State whether each of the following is a good or poor conductor of electricity in the solid state:

a. Na₂SO₄b. Xec. SiCd. Zn

3

Rank the following from lowest to highest boiling point:

 CH_3NH_2 , CO, H_2 , N_2 , SiO_2

4

Predict whether each solute below will dissolve to a greater extent in carbon tetrachloride or water:

- a. H_2O_2
- b. Br_2
- c. HCN
- d. NH_4NO_3



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