# Non Sibi High School

#### Andover's Chem 250: Introductory/Basic Chemistry

#### Chapter 12, Review Quiz 1

## 1

Rank the compounds  $CaBr_2$ , KI, MgS, and  $SiCl_4$  from lowest to highest melting point.

### $\mathbf{2}$

State whether each of the following is a good or poor conductor of electricity in the solid state:

a. Na<sub>2</sub>SO<sub>4</sub>b. Xec. SiCd. Zn

# 3

Rank the following from lowest to highest boiling point:

 $CH_3NH_2$ , CO,  $H_2$ ,  $N_2$ ,  $SiO_2$ 

#### 4

Predict whether each solute below will dissolve to a greater extent in carbon tetrachloride or water:

- a.  $H_2O_2$
- b.  $Br_2$
- c. HCN
- d.  $NH_4NO_3$



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