Non Sibi High School

Andover's Chem 250: Introductory/Basic Chemistry Chapter 9, Review Quiz 1

1

Write the chemical formula for each compound:

- a. ammonium phosphate
- b. butanol
- c. aluminum sulfate
- d. potassium nitride

$\mathbf{2}$

Write the name of each compound:

- a. $Fe(OH)_3$
- b. P_2O_5
- c. $Mg(HCO_3)_2$
- $\mathrm{d.}\ C_6\mathrm{H}_{14}$

3

Draw Lewis structures for:

- a. boron trichloride
- b. SF $_5$ $^-$

4

Draw Lewis structures for H₂CNH and CH₃NH₂. Which molecule will have the longer carbon-to-nitrogen bond length?

5

Draw all the resonance structures for NO $_2\,^-$



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